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Protecting group free transformations of reducing sugars in aqueous solution

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The development of new selective reactions of completely unprotected sugars, particularly in aqueous solution, has become an area of resurgent interest. Seminal studies reported by Shoda¹ and co-workers first introduced the dehydrating reagent 2-chloro-1,3-dimethylimidazolinium chloride (DMC) into the carbohydrate field, and revealed its remarkable ability to selectively activate the anomeric hydroxyl group of unprotected sugars in aqueous solution. A series of highly useful protecting group-free processes based on the use of DMC and analogues has since been developed.² I will discuss some of our recent work in this area,³ building on the important developments of Shoda and others, focussing on the application of DMC and analogues for the direct conversion of unprotected sugars into a range of glycosides, glycoconjugates, and even (1-6)-linked disaccharides⁴ without the need for any protecting group chemistry.

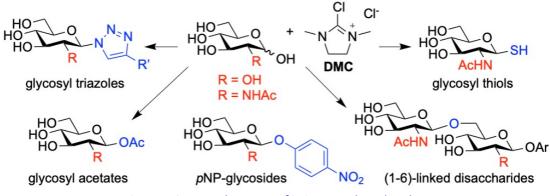
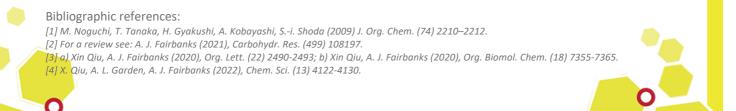


Figure 1: Structural concept of iminosugar based probes.



Green (glyco)chemistry and sustainable development / New reactions involving sugars and mimetics